The authors thank D. Heijdenrijk for collecting the X-ray data.

Lists of structure factors, anisotropic thermal parameters, H-atom coordinates and complete geometry have been deposited with the British Library Document Supply Centre as Supplementary Publication No. SUP 71304 (14 pp.). Copies may be obtained through The Technical Editor, International Union of Crystallography, 5 Abbey Square, Chester CH1 2HU, England. [CIF reference: SH1048]

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Structure of N,N'-Dimethylpiperazine-2,3-dithione: Space Group Correction

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Abstract

The crystal structure of N,N'-dimethylpiperazine-2,3-dithione, $C_6H_{10}N_2S_2$, has been described and refined in space group Aa [Servaas, Stufkens, Oskam, Vernooijs, Baerends, De Ridder & Stam (1989). *Inorg. Chem.* 28,

4104-4113]. It is described properly in space group A2/a. Revised coordinates and bond lengths and angles are given.

Comment

The structure of the title compound was reported in space group Aa (Servaas $et\ al.$, 1989). The six-membered ring of the molecule was found to adopt a half-chair conformation with the twofold axis running through the midpoints of the C—C bonds, parallel to the b axis. Furthermore, the two N—C(methyl) bond lengths were very dissimilar [1.439 (5) and 1.489 (5) Å]; there is no reason for them to be unequal. This unusual spread of distances undoubtedly resulted from the refinement of a centrosymmetric structure in a non-centrosymmetric space group (Ermer & Dunitz, 1970; Schomaker & Marsh, 1979). Therefore it seemed likely that the structure should properly be described in space group A2/a.

Starting coordinates were from Table 3 of Servaas *et al.* (1989) and averaged in accordance with A2/a. Full-matrix minimization of the quantity $\Sigma w(F_o - F_c)^2$ with $w = 1/[\sigma^2(F) + 0.0035F^2]$ was performed. Refinement was anisotropic for the non-H atoms and isotropic for the H atoms. An isotropic extinction coefficient was included in the parameters [Zachariasen, final value $3.4 (5) \times 10^{-5}$]. Anomalous dispersion was taken into account but no correction for absorption was applied. The H-atom positions were calculated initially on the basis of standard geometry and refined. A final difference Fourier map revealed residual electron density within the range $-0.22-0.30 \, \mathrm{e} \, \mathrm{A}^{-3}$.

The values obtained for parameters, bond lengths and angles are close to the averages of pairs of equivalent values obtained in the earlier Aa refinement. The two N—C(methyl) bond lengths are now equal by symmetry at 1.473 (3) Å. The general description of the structure remains unchanged.

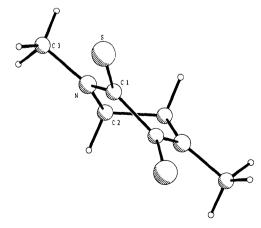


Fig. 1. A *PLUTO* (Motherwell & Clegg, 1978) drawing of C₆H₁₀N₂S₂ showing the labelling of the independent non-H atoms. The view is down the twofold axis, with the crystallographic 2₁ axis vertical.

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Experimental

Crystal data	
$C_6H_{10}N_2S_2$ $M_r = 174.29$ Monoclinic $A2/a$ $a = 10.653 (1) Å$ $b = 7.2936 (6) Å$ $c = 10.6398 (9) Å$ $\beta = 95.26 (1)^\circ$ $V = 823.2 (1) Å^3$	D_x = 1.406 Mg m ⁻³ Cu $K\alpha$ radiation λ = 1.5418 Å Cell parameters from 23 reflections θ = 38-44° μ = 5.20 mm ⁻¹ T = 293 K Needle shaped crystals 0.25 × 0.20 × 0.13 mm Dark red

Data collection

Duit concern.	
Enraf-Nonius CAD-4	$\theta_{\rm max}$ = 69.71°
diffractometer	$h = -12 \rightarrow 12$
$\theta/2\theta$ scans	$k = 0 \rightarrow 8$
Absorption correction:	$l = 0 \rightarrow 12$
none	2 standard reflections
1710 measured reflections	frequency: 66.67 min
770 independent reflections	intensity variation: none
731 observed reflections	
$[I > 2.5\sigma(I)]$	

Refinement

Refinement on F	Extinction correction:
Final $R = 0.037$	Zachariasen (1967)
wR = 0.07	Extinction coefficient:
S = 1.167	$g = 3.4(5) \times 10^{-5}$
714 reflections	Atomic scattering factors
67 parameters	from International Tables
refined	for X-ray Crystallogra-
$w = 1/[\sigma^2(F) + 0.0035F^2]$	phy (1974, Vol. IV, Tables
$(\Delta/\sigma)_{\rm max} = 0.783$	2.2B and 2.3.1)
$\Delta \rho_{\text{max}} = 0.294 \text{ e Å}^{-3}$	
$\Delta \rho_{\min} = -0.23 \text{ e Å}^{-3}$	

Table 1. Fractional atomic coordinates and equivalent isotropic thermal parameters (Å²)

The origin is located at a centre of symmetry on the glide plane a. $U_{\text{eq}} = \frac{1}{3} \sum_i \sum_j U_{ij} a_i^* a_i^* \mathbf{a}_i . \mathbf{a}_j.$

	x	y	z	$U_{ m eq}$
S	0.11142 (4)	0.18040 (6)	0.05913 (4)	0.0460 (6)
C1	0.1977 (1)	0.3686(3)	0.0442 (1)	0.0331 (9)
C2	0.2460(3)	0.6922(3)	0.0694(3)	0.057(1)
C3	0.0912 (2)	0.5405 (4)	0.2013 (2)	0.058(1)
N	0.1807 (1)	0.5247 (2)	0.1041 (1)	0.0425 (9)

Table 2. Geometric parameters (Å, °)

S—C1	1.668 (2)	C2-N	1.470 (3)		
C1-N	1.325 (2)	C2—C2i	1.487 (4)		
C1—C1 ⁱ	1.523 (2)	C3—N	1.473 (3)		
SC1N	124.1 (1)	C1-N-C2	120.3 (2)		
S-C1-C1 ⁱ	120.0(1)	C1—N—C3	122.0 (2)		
N-C1-C1i	115.9 (2)	C2—N—C3	117.6 (2)		
NC2C2i	108.7 (2)				
Symmetry code: (i) $\frac{1}{2} - x, y, -z$.					

Crystals were prepared by Dr P. C. Servaas according to the literature procedure described by Isaksson, Liljefors & Sandström (1981). Data collection: *CAD-4 Software* (Enraf-Nonius, 1989).

Cell refinement: CELCON program comparable to Xtal LAT-CON (Hall, Flack & Stewart, 1992). Data reduction: Xtal AD-DREF, SORTRF. Structure solved by the heavy-atom method. Program(s) used to refine structure: Xtal CRYLSQ. Molecular graphics: PLUTO (Motherwell & Clegg, 1978). Software used to prepare material for publication: Xtal BONDLA, CIFIO.

Lists of structure factors, anisotropic thermal parameters, H-atom coordinates and complete geometry have been deposited with the British Library Document Supply Centre as Supplementary Publication No. SUP 71346 (8 pp.). Copies may be obtained through The Technical Editor, International Union of Crystallography, 5 Abbey Square, Chester CH1 2HU, England. [CIF reference: SH1061]

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Structure of Dimethyl 2-[o-(3,5-Dimethyl-1-pyrazolyl)anilino]-3-methoxymaleate

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(Received 2 April 1993; accepted 28 May 1993)

Abstract

In the title compound the pyrazole and benzene rings form a dihedral angle of 64.5 (1)°. Neither ring deviates significantly from planarity. There is an intramolecular hydrogen bond between atom N12 and atom O23 [N12···O23 2.682 (4) Å, N12—H12···O23 102.9 (2)°]; all other bond distances and angles are within the expected ranges.

Comment

This work is part of a more complex study on the chemistry of the Ramsden's Class B heteropentalenes. The ti-